Received 20 May 2005

Accepted 23 May 2005

Online 28 May 2005

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Gaduo, Jian-Rong Su and Duan-Jun Xu*

Department of Chemistry, Zhejiang University, Hangzhou, 310027, People's Republic of China

Correspondence e-mail: chem@zju.edu.cn

Key indicators

Single-crystal X-ray study T = 295 K Mean σ (C–C) = 0.003 Å R factor = 0.021 wR factor = 0.053 Data-to-parameter ratio = 16.9

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e. Benzimidazolium bis(1*H*-benzimidazole- κN^3)chloro(oxydiacetato- $\kappa^3 O, O', O''$)cadmium(II)

The title compound, $(C_7H_7N_2)[Cd(C_4H_4O_5)Cl(C_7H_6N_2)_2]$, is composed of Cd^{II}-containing complex anions and benzimidazolium (HBZIM) cations. The Cd^{II} atom is coordinated by a tridentate oxydiacetate (ODA) dianion, a chloride anion and two benzimidazole (BZIM) molecules, forming a distorted octahedral CdN₂O₃Cl coordination geometry. The ODA dianion chelates facially to the Cd^{II} atom. The face-to-face distances of 3.444 (12) and 3.386 (5) Å between partially overlapped BZIM rings and between HBZIM rings, respectively, indicate the existence of π - π stacking in the crystal structure.

Comment

As part of our ongoing investigations of π - π stacking in metal complexes incorporating benzimidazole (BZIM) ligands (Su *et al.*, 2005), the title Cd^{II} complex, (I) (Fig. 1), has been prepared and its crystal structure is presented here.



The Cd^{II}-containing complex anion displays a distorted octahedral CdN₂O₃Cl coordination geometry, formed by a tridentate oxydiacetate (ODA) dianion, a chloride anion and two BZIM molecules. The ODA dianion chelates to the Cd^{II} atom in a facial configuration, the two five-membered chelate rings being nearly perpendicular to each other (Fig. 1). The O1-ring displays an envelope configuration, with atom O3 at the flap position, displaced by 0.239 (2) Å from the mean plane of the other atoms. Conversely, the O4-ring is twisted about the Cd–O3 bond, and the Cd atom lies 0.832 (6) Å out of the plane formed by atoms C3, C4 and O4, while atom O3 is displaced by -0.341 (5) Å with respect to the same three atoms.

The molecular packing is illustrated in Fig. 2, showing the parallel arrangement of BZIM molecules and benzimidazolium (HBZIM) cations in the crystal structure of (I). The parallel N33-HBZIM and N33^v-HBZIM [symmetry code: (v) -x, -y, 1 - z] cations overlap their imidazole rings, whereas

Printed in Great Britain – all rights reserved Acta Cryst. (2005). E61, m1231–m1233

© 2005 International Union of Crystallography



Figure 1

The molecular structure of (I), with 30% probability displacement ellipsoids (spheres of arbitrary radii for H atoms). The dashed line indicates a hydrogen-bonding interaction.



Figure 2

The packing in (I), showing the hydrogen bonding (dashed lines). [Symmetry codes: (ii) 1 - x, 1 - y, -z; (iv) x, 1 + y, z.]



A diagram showing π - π stacking between HBZIM cations. [Symmetry code: (v) -x, -y, 1 - z.]

the parallel C19-BZIM and C19^{vi}-BZIM [symmetry code: (vi) 2 - x, 1 - y, -z] molecules overlap their benzene rings, as shown in Figs. 3 and 4. The face-to-face distances of 3.386 (5) and 3.444 (12) Å between imidazole rings and between benzene rings, respectively, reflect the π - π stacking interactions in the crystal structure of (I).

Neighbouring Cd^{II} complex anions are linked through N– H···O hydrogen bonds. N–H···O hydrogen-bond interactions are also observed between Cd^{II} complex anions and HBZIM cations (Table 2 and Fig. 2).

Experimental

All starting chemicals were of analytical grade and were used as received. H_2ODA (0.13 g, 1 mmol), Na_2CO_3 (0.11 g, 1 mmol), BZIM (0.24 g, 2 mmol) and CdCl₂·2.5H₂O (0.23 g, 1 mmol) were dissolved in a water–ethanol mixture (20 ml, 3:1). The solution was refluxed for 4 h, cooled to room temperature and filtered. The filtrate was kept at room temperature. Colourless single crystals of (I) were obtained from the filtrate after about one week.

Crystal data

$(C_7H_7N_2)[Cd(C_4H_4O_5)Cl-$	Z = 2
$(C_7H_6N_2)_2]$	$D_x = 1.640 \text{ Mg m}^{-3}$
$M_r = 635.34$	Mo $K\alpha$ radiation
Triclinic, P1	Cell parameters from 8625
a = 9.016 (2) Å	reflections
b = 11.1672 (14) Å	$\theta = 2.0-26.0^{\circ}$
c = 13.243 (3) Å	$\mu = 1.00 \text{ mm}^{-1}$
$\alpha = 78.47 \ (1)^{\circ}$	T = 295 (2) K
$\beta = 80.58 \ (3)^{\circ}$	Block, colourless
$\gamma = 84.82 \ (1)^{\circ}$	$0.40 \times 0.30 \times 0.28 \text{ mm}$
$V = 1286.5 (5) \text{ Å}^3$	

Data collection

Rigaku R-AXIS RAPID diffractometer ω scans Absorption correction: multi-scan (*ABSCOR*; Higashi, 1995) $T_{min} = 0.670, T_{max} = 0.754$ 12674 measured reflections

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.021$ $wR(F^2) = 0.053$ S = 1.085825 reflections 344 parameters H-atom parameters constrained 5825 independent reflections 5419 reflections with $I > 2\sigma(I)$ $R_{int} = 0.020$ $\theta_{max} = 27.5^{\circ}$ $h = -10 \rightarrow 11$ $k = -14 \rightarrow 14$ $l = -17 \rightarrow 17$

$$\begin{split} &w = 1/[\sigma^2(F_o^2) + (0.0336P)^2 \\ &+ 1.1911P] \\ &where \ P = (F_o^2 + 2F_o^2)/3 \\ (\Delta/\sigma)_{max} = 0.003 \\ \Delta\rho_{max} = 0.31 \ e^{A^{-3}} \\ \Delta\rho_{min} = -0.38 \ e^{A^{-3}} \\ Extinction \ correction: \ SHELXL97 \\ Extinction \ coefficient: \ 0.0114 \ (5) \end{split}$$

Table 1	
Selected geometric p	oarameters (Å, °).

Cd-O1	2.2909 (12)	Cd-N13	2.3235 (15)
Cd-O3	2.4975 (13)	Cd-N23	2.2708 (15)
Cd-O4	2.3727 (13)	Cd-Cl	2.5213 (5)
O1-Cd-O3	68.92 (4)	O3-Cd-Cl	91.61 (3)
O1-Cd-O4	88.49 (4)	O4-Cd-N13	171.42 (5)
O1-Cd-N13	89.05 (5)	O4-Cd-N23	94.40 (5)
O1-Cd-N23	95.46 (5)	O4-Cd-Cl	86.95 (4)
O1-Cd-Cl	160.20 (3)	N13-Cd-N23	94.01 (6)
O3-Cd-O4	66.84 (5)	N13-Cd-Cl	92.62 (4)
O3-Cd-N13	104.63 (5)	N23-Cd-Cl	104.10 (4)
O3-Cd-N23	155.08 (5)		

Table 2

Hydrogen-bond geometry (Å, $^{\circ}$).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots \mathbf{A}$
N11-H11···O4 ⁱ	0.86	1.97	2.769 (2)	155
$N21 - H21 \cdot \cdot \cdot O2^{ii}$	0.86	1.95	2.786 (2)	165
$N31 - H31 \cdot \cdot \cdot O5^{iii}$	0.86	1.83	2.670 (2)	164
N33-H33···O2	0.86	1.83	2.673 (2)	168

Symmetry codes: (i) x + 1, y, z; (ii) -x + 1, -y + 1, -z; (iii) x, y - 1, z.

H atoms were positioned geometrically, with C-H = 0.93 (aromatic) or 0.97 Å (methylene) and N-H = 0.86 Å, and included in the final cycles of refinement in the riding model, with the constraint $U_{\rm iso}({\rm H}) = 1.2U_{\rm eq}({\rm carrier})$ applied.

Data collection: *PROCESS-AUTO* (Rigaku Corporation, 1998); cell refinement: *PROCESS-AUTO*; data reduction: *CrystalStructure* (Rigaku/MSC & Rigaku Corporation, 2002); program(s) used to



Figure 4

A diagram showing π - π stacking between BZIM molecules of neighboring Cd^{II} complex anions. [Symmetry code: (vi) 2 - x, 1 - y, -z.]

solve structure: *SIR92* (Altomare *et al.*, 1993); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

This project was supported by the National Natural Science Foundation of China (grant No. 20443003).

References

- Altomare, A., Cascarano, G., Giacovazzo, C. & Guagliardi, A. (1993). J. Appl. Cryst. 26, 343–350.
- Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.
- Farrugia, L. J. (1999). J. Appl. Cryst. 32, 837-838.
- Higashi,, T. (1995). ABSCOR. Rigaku Corporation, Tokyo, Japan.
- Rigaku Corporation (1998). PROCESS-AUTO. Rigaku Corporation, Tokyo, Japan.
- Rigaku/MSC & Rigaku Corporation (2002). CrystalStructure. Version 3.00. Rigaku/MSC, The Woodlands, TX 77381-5209, USA, and Rigaku Corporation, Akishima, Tokyo, Japan.
- Sheldrick, G. M. (1997). SHELXL97. University of Göttingen, Germany.
- Su, J.-R., Gu, J.-M. & Xu, D.-J. (2005). Acta Cryst. E61, m244-m246.